



An unexpected retro-aldol–aldol in the AB-ring in the synthesis of (\pm)-arisugacin A

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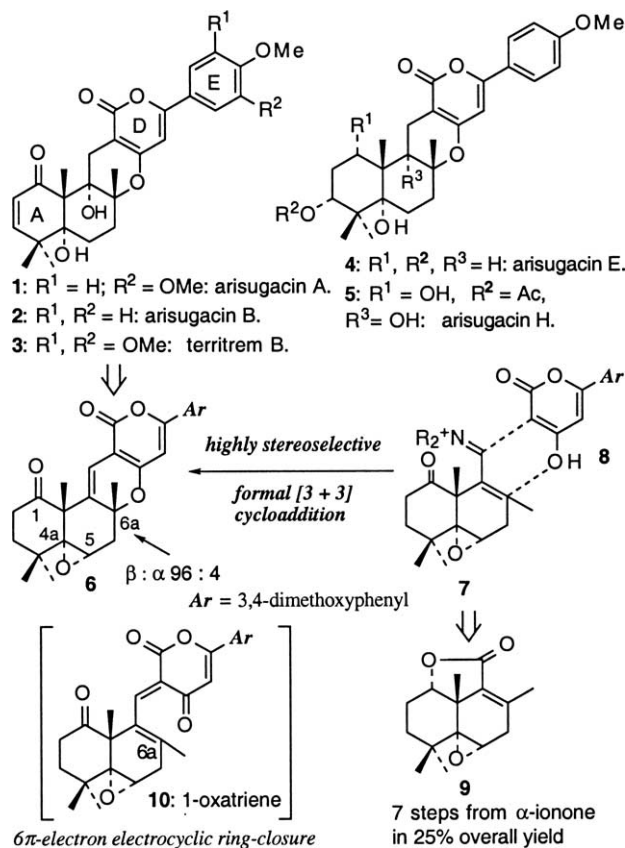
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Abstract—Endeavors including an unexpected retro-aldol–aldol process that proved to be critical for our eventual total synthesis of arisugacin A are described here. © 2002 Elsevier Science Ltd. All rights reserved.

Arisugacin A (**1**), isolated from *Penicillium* sp. Fo-4259 by Ōmura, is the most potent and selective inhibitor known against acetylcholinesterase with an IC_{50} of 1 nM,¹ thereby possessing significance in the treatment of dementias.² Given its biological relevance and unique structural features where it resembles other important natural products such as the territrems (**3**)³ and pyripyropenes,⁴ we have explored a number of different synthetic routes seeking an efficient synthesis of arisugacin A.^{5–8} More than 2 years ago, we reported the first synthesis of the advanced pentacyclic intermediate **6**^{7,8} via a highly stereoselective formal [3+3] cycloaddition reaction^{9–13} of α,β -unsaturated iminium salt **7** with 2-pyrone **8**. This proceeds through a stereoselective 6π -electron electrocyclic ring-closure¹⁴ of the 1-oxatriene **10** (Scheme 1). Such a tandem Knoevenagel-pericyclic ring-closure sequence¹⁵ furnishes two σ -bonds and a new stereocenter adjacent to the heteroatom (C6a in **6**). We have since developed this stepwise cycloaddition into a useful method for constructing complex 2*H*-pyrans¹⁰ and dihydropyridines.¹² Recently, Ōmura and Sunazuka reported an identical route to **6**,¹⁶ thereby supporting the high degree of efficiency and convergency of our approach and validating the synthetic potential of this formal [3+3] cycloaddition strategy.

Although the 13-step preparation of the advanced pentacycle **6** via the formal [3+3] cycloaddition from α -ionone represents success in meeting a significant challenge, there remained several unexpected challenges in the synthesis of arisugacin A. The foremost problem

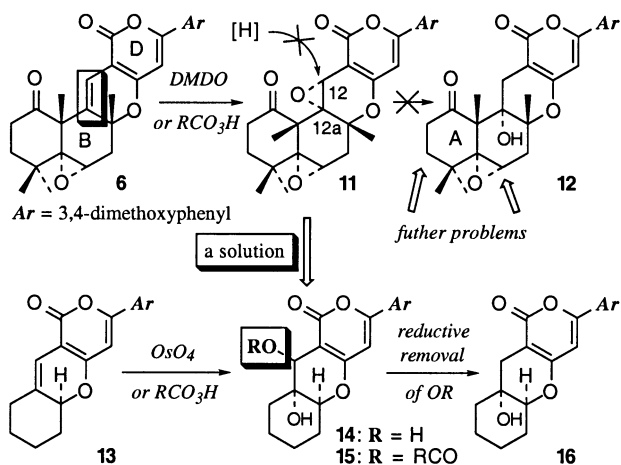
was the installation of the C12a hydroxyl functionality via a sequence of epoxidation of the C-ring olefin in **6** followed by reduction of **11** (Scheme 2). This epoxidation–reduction sequence was difficult because rapid



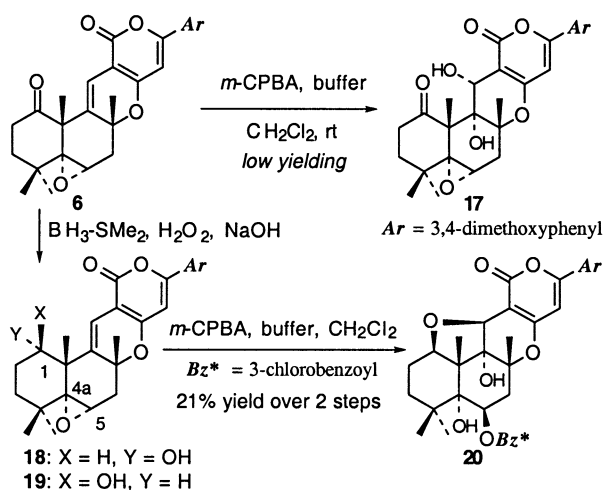
Scheme 1.

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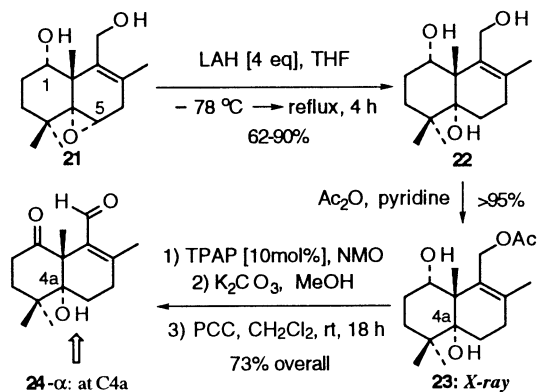
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Scheme 2.



Scheme 3.



Scheme 4.

ring-opening of the C12a-C12 epoxide in **11** by various nucleophilic oxygen species precluded the addition of hydride.

Our solution ultimately involved bisoxygenation of **13** followed by removal of the activated secondary oxygen functionality in **14** or **15** using reductive methods (Scheme 2).¹⁷ Omura and Sunazuka encountered a sim-

ilar problem and employed our protocols¹⁷ in their recent total synthesis of (±)-arisugacin A.¹⁸ In our own effort, we met with further obstacles associated with ring-opening of the C4a-C5 epoxide and construction of the enone A-ring. We communicate here our endeavors in meeting these challenges and an unexpected retro-aldol-aldol in the AB-ring.

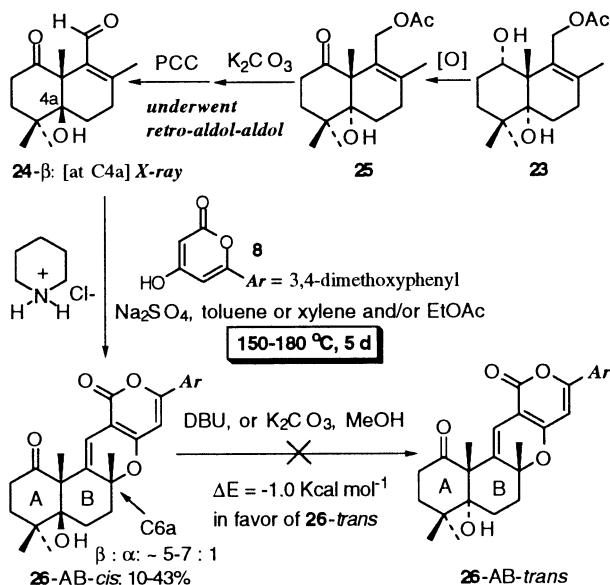
Based on our model study (**13** to **16**),¹⁷ the epoxy pentacycle **6** was subjected to a variety of epoxidation and dihydroxylation conditions, but the only discernable product arising from these oxidative attempts was **17**¹⁹ in only 8–21% yield when using *m*-CPBA (Scheme 3). Attempts to hydroborate the same olefin instead led to the exclusive formation of alcohol **19** with β-C1-OH (based on NOE). Interestingly, reduction of the same ketone using Dibal-H or NaBH₄ led to the same stereochemical outcome.

The ensuing epoxidation of **19** led to the hexacycle **20** in 21% overall yield from **6**. The stereochemistry of **20** was assigned at this point using NOE experiments. While epoxidation of the C-ring olefin had taken place, an intramolecular ring-opening of the epoxide by the β-C1-OH also occurred to give the furan. Surprisingly, ring-opening of the C4a-C5 epoxide in the B-ring had also occurred. After subsequent Dibal-H reduction (63% yield) and xanthate formation (CS₂, MeI, 52% yield), Barton's deoxygenation proceeded only in low yield, thereby stifling this synthetic route.

To avoid the C4a-C5 epoxide opening issue, we prepared the keto enal **24-α** in six steps from epoxy diol **21** (Scheme 4).^{7,20,21} The key triol **22** was reported to be obtainable directly from epoxy lactone **9** via a global reduction using LAH and AlCl₃ reduction.¹⁶ However, to obtain **22** effectively and consistently, a stepwise reduction from **9** was needed. That is treatment of **9** with LAH to give **21** followed by its treatment with LAH at 65°C in THF.

In the ensuing steps, we did not immediately detect anything unusual regarding the keto enal **24-α**, especially since we had an X-ray structure of **23** reaffirming all relative stereochemistry. However, it was found that the subsequent formal [3+3] cycloaddition of the alleged keto enal **24-α** with **8**²² could only proceed at high temperatures in low yields with relatively poor diastereoselectivity despite using the more reactive piperidinium hydrochloride salt²³ (Scheme 5). This outcome was surprising given our experiences with this reaction and prompted us to carefully examine various intermediates.

The X-ray structure of **24** disturbingly showed that we had not prepared the expected keto enal **24-α** but instead we had **24-β**. It can be quickly rationalized that a *retro-aldol-aldol* sequence had likely occurred during preparations of **24** during deacylation of **25**. Thus, the key cycloaddition had led to the unexpected pentacycle **26-cis** (AB-ring). Calculations (AMI-Spartan™) intriguingly showed that **24-β** is actually more stable than **24-α** by ca. 1.5 kcal mol⁻¹. Presumably, **24-β**



Scheme 5.

alleviates 1,3-diaxial interactions between the methyl groups in **24-α** and serves as a driving force for this unexpected *retro-aldol-aldol* sequence. However, despite the fact that the desired pentacycle **26-trans** is more stable than **26-cis** by ~ 1.0 kcal mol⁻¹, all attempts to isomerize **26-cis** to **26-trans** failed.

In summary, we have described here our efforts towards synthesis of arisugacin A and have found difficulties in the transformation of the C-ring olefin to the desired C12a hydroxyl group and the ring-opening of the B-ring epoxide despite success in our model studies. An alternative approach involving construction of the pentacycle **26** led to an unexpected *retro-aldol-aldol* process that occurred during the constructions of the AB ring. These endeavors proved to be critical for our eventual total synthesis of arisugacin.

Acknowledgements

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23. These reactions are best carried out in toluene and/or EtOAc using piperidine/Ac₂O, or piperidinium salts as well as other amine salts. Hydrochloride salts tend to be even more reactive. L-Proline does not work well for this reaction contrary to other literature accounts.
24. Selected characterizations: **17**: $R_f=0.46$ (65% EtOAc in hexane); ^1H NMR (300 MHz, CDCl₃) δ 0.88 (s, 3H), 1.18 (s, 3H), 1.25 (m, 1H), 1.48 (s, 3H), 1.54 (s, 3H), 1.70 (m, 1H), 2.01 (m, 1H), 2.43 (d, 1H, $J=16.8$ Hz), 2.81 (ddd, 1H, $J=7.5, 11.7, 19.2$ Hz), 2.99 (dd, 1H, $J=3.9, 16.8$ Hz), 3.48 (d, 1H, $J=3.9$ Hz), 3.92 (s, 3H), 3.94 (s, 3H), 3.95 (s, 1H), 6.11 (s, 1H), 6.92 (d, 1H, $J=8.6$ Hz), 7.36 (d, 1H, $J=2.1$ Hz), 7.53 (dd, 1H, $J=2.1, 8.6$ Hz); mass spectrum (EI): m/e (%relative intensity) 513 (10) M^+ +H, 447 (7), 391 (14), 307 (73), 154 (100), 107 (74); m/e calcd for C₂₈H₃₃O₉: 513.2125, found: 513.2130. **20**: $R_f=0.22$ (66% EtOAc in hexane); ^1H NMR (500 MHz, CDCl₃) δ 1.00 (s, 3H), 1.03 (s, 3H), 1.17–1.24 (m, 1H), 1.71–1.77 (m, 1H), 1.75 (s, 3H), 1.77 (s, 3H), 1.81–1.87 (m, 1H), 1.96 (d, 1H, $J=15.5$ Hz), 2.11 (dt, 1H, $J=2.5, 13.5$ Hz), 2.38 (d, 1H, $J=4.0$ Hz), 2.67 (dd, 1H, $J=7.0, 15.5$ Hz), 3.69 (s, 1H), 3.94 (s, 3H), 3.96 (s, 3H), 4.66 (brd, 1H, $J=11.0$ Hz), 5.00 (s, 1H), 5.55 (d, 1H, $J=7.0$ Hz), 6.33 (s, 1H), 6.91 (d, 1H, $J=8.5$ Hz), 7.30 (d, 1H, $J=2.0$ Hz), 7.40 (dd, 1H, $J=2.0, 8.5$ Hz), 7.45 (t, 1H, $J=8.0$ Hz), 7.59 (ddd, 1H, $J=1, 2.0, 8.0$ Hz), 7.87 (td, 1H, $J=2.0, 8.0$ Hz), 7.97 (t, 1H, $J=2.0$ Hz); mass spectrum (CI): m/e (%relative intensity) 653 ($M+H^+$) (100), 315 (15), 277 (16), 261 (13), 203 (12), 165 (48), 139 (89), 119 (36); m/e calcd for C₃₅H₃₈O₁₀Cl: 653.2154; found: 653.215. **22**: mp 135.5–137.5°C; $R_f=0.42$ (25% acetone in chloroform); ^1H NMR (300 MHz, CDCl₃) δ 1.03 (s, 3H), 1.05 (s, 3H), 1.14 (m, 1H), 1.26 (m, 1H), 1.68–1.88 (m, 3H), 2.00–2.14 (m, 3H), 2.18–2.30 (m, 1H), 3.46 (s, 1H), 4.00 (m, 1H), 4.06 (d, 1H, $J=11.7$ Hz), 4.07 (brs, 1H), 4.31 (dd, 1H, $J=7.2, 11.7$ Hz), 5.23 (d, 1H, $J=7.2$ Hz); ^{13}C NMR (75 MHz, CDCl₃) δ 19.2, 23.5, 24.6, 25.3, 25.7, 27.7, 28.7, 30.2, 38.0, 46.3, 57.4, 73.4, 78.0, 134.2, 135.5; IR (film) cm⁻¹ 3356 brs, 2966 s, 1654 m; mass spectrum (EI): m/e (%relative intensity) 236 ($M-H_2O^+$) (35), 221 (40), 185 (33), 162 (55), 152 (47), 139 (61), 137 (69), 119 (61), 109 (66), 81 (100). **24-β**: mp 162–164°C; $R_f=0.21$ (50 EtOAc in hexanes); ^1H NMR (500 MHz, CDCl₃) δ 1.04 (s, 3H), 1.16 (s, 3H), 1.17 (s, 3H), 1.42 (s, 1H), 1.52 (ddd, 1H, $J=5.0, 5.0, 14.0$ Hz), 1.77 (ddd, 1H, $J=8.0, 9.5, 18.0$ Hz), 1.95 (ddd, 1H, $J=4.0, 13.5, 13.5$ Hz), 2.12 (ddd, 1H, $J=2.0, 8.0, 14.5$ Hz), 2.18 (s, 3H), 2.34 (ddd, 1H, $J=4.0, 4.0, 12.0$ Hz), 2.46–2.63 (m, 3H), 10.01 (s, 1H); ^{13}C NMR (125 MHz, CDCl₃) δ 18.5, 20.2, 24.7, 26.5, 27.8, 34.5, 37.8, 39.4, 39.8, 54.0, 81.3, 139.2, 158.4, 189.9, 212.7; IR (film) cm⁻¹ 3488 br, 2950 m, 1707 vs, 1665 vs; mass spectrum (EI): m/e (%relative intensity) 250 M^+ (37), 217 (24), 181 (36), 152 (100), 138 (40), 123 (49), 109 (34), 55 (44). **26-AB-cis**: mp 188–190°C; $R_f=0.39$ (67% EtOAc in hexanes); ^1H NMR (500 MHz, CDCl₃) δ 0.91 (s, 3H), 1.10 (s, 3H), 1.31 (s, 3H), 1.45 (s, 3H), 1.49 (s, 1H), 1.61–1.73 (m, 3H), 2.04 (ddd, 1H, $J=4.5, 8.5, 14.0$ Hz), 2.13 (m, 1H), 2.23 (ddd, 1H, $J=4.5, 9.5, 18.5$ Hz), 2.40 (ddd, 1H, $J=7.0, 7.0, 17.0$ Hz), 2.64 (ddd, 1H, $J=7.5, 7.5, 17.0$ Hz), 3.87 (s, 3H), 3.88 (s, 3H), 6.01 (s, 1H), 6.33 (s, 1H), 6.84 (d, 1H, $J=8.5$ Hz), 7.24 (d, 1H, $J=2.5$ Hz), 7.34 (dd, 1H, $J=2.5, 9.0$ Hz); ^{13}C NMR (125 MHz, CDCl₃) δ 23.0, 24.0, 26.1, 26.9, 30.2, 33.2, 33.4, 35.9, 38.9, 56.1, 56.2, 58.1, 78.9, 80.7, 96.1, 100.9, 108.3, 111.1, 114.3, 119.2, 124.0, 135.8, 149.3, 151.6, 160.6, 161.4, 163.0, 212.0; IR (film) cm⁻¹ 3484 br, 2960 m, 1704 vs, 1625 m, 1515 vs; mass spectrum (CI): m/e (%relative intensity) 480 (67), 465 (22), 446 (25), 437 (13), 327 (49), 261 (24), 165 (100), 137 (12); m/e calcd for C₂₈H₃₂O₇: 480.2148; found: 480.2143.